



CO From CO₂ via Chemical Looping Dominick Battaglia, Jace Carsten, William McPhaul, Zacharia Nyambega Mentors: Dr. Luke Neal

1. Motivation

- Addressing the climate crisis by using waste gas sources for CO production
- Conversion of CH₄ and CO₂ to CO reduces greenhouse gas presence
- CO and H₂ produced from the redox process have various uses in industry:
 - Reducing agent, cleaning detergent, methanol production, food packaging

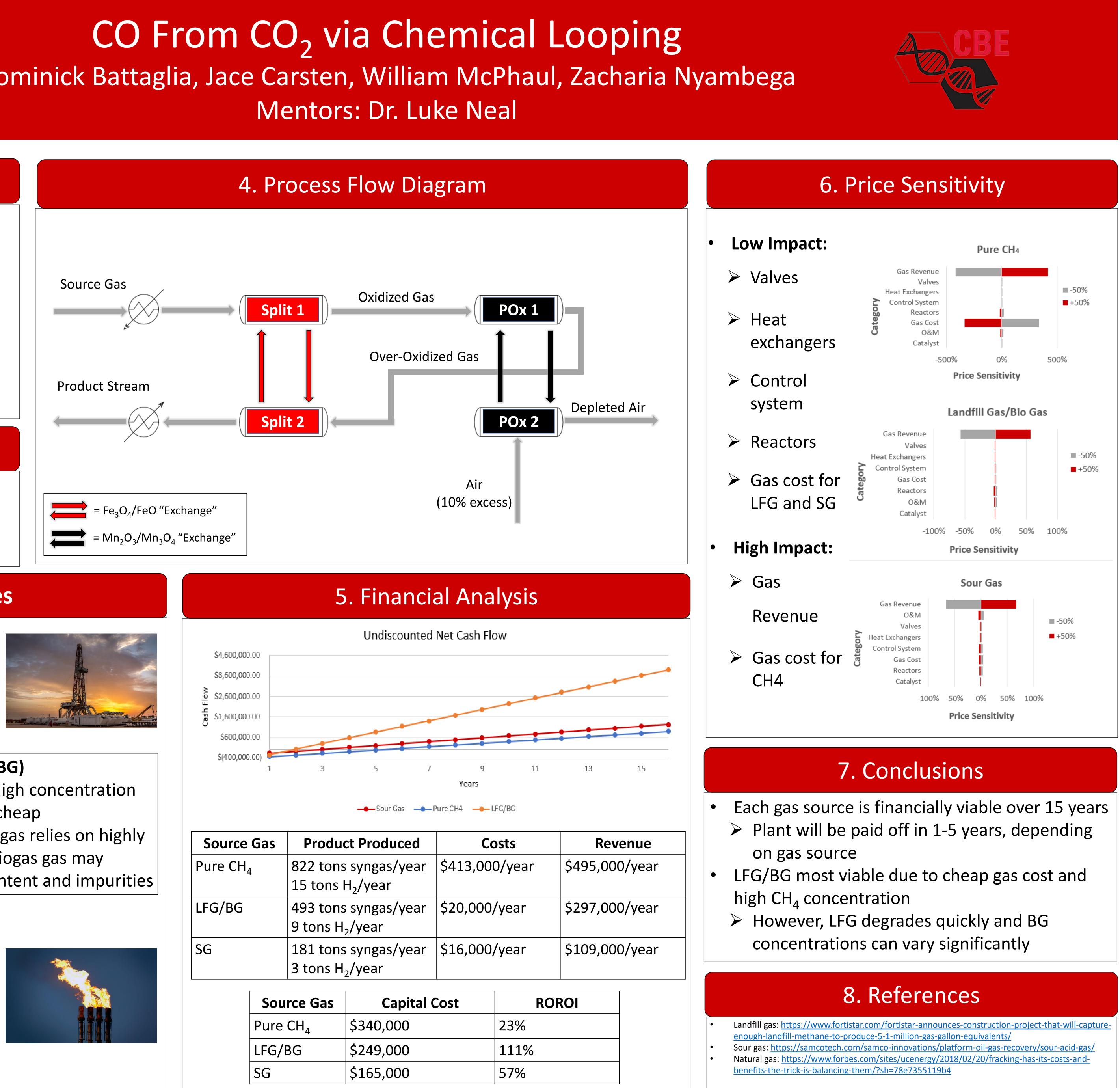
2. Goals

- Enhance environmental sustainability
- Determine engineering and economic viability of the process
- Determine best case gas source

3. Gas Sources

Natural Gas (Pure CH₄)

- Upsides: high concentration of CH₄
- Downsides: Expensive, non-renewable, fracking





Landfill / Biogas (LFG/BG)

- Upsides: relatively high concentration of CH_4 , renewable, cheap
- Downsides: Landfill gas relies on highly polluting landfills; biogas gas may contain high CO₂ content and impurities

Sour Gas (SG)

- Upsides: cheap, natural gas collection byproduct
- Downsides: High H₂S content, low CH₄ content, source is underdeveloped

